

# Thermo & Quantum Mechanics

# Thermodynamics Equations

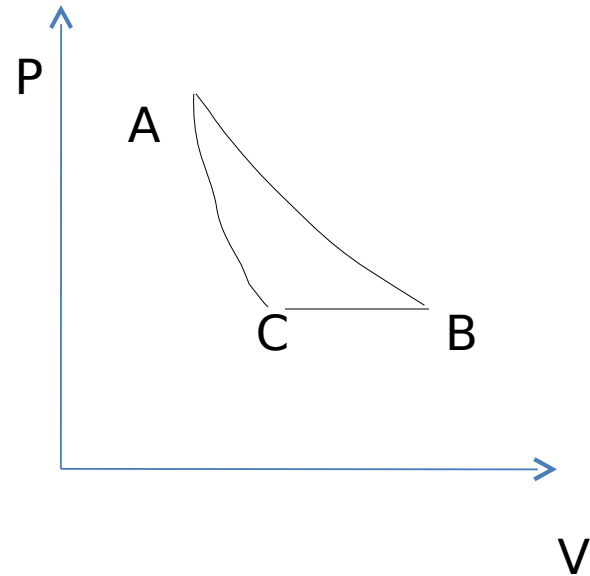
- $\Delta U = Q + W$                       1st Law
- $W = -\int P dV$
- $dQ = T dS$                       for reversible processes
- $\Delta U = m C_v \Delta T$
- $\Delta S_{\text{universe}} \geq 0$               2nd Law
- $H = U + PV$                       Enthalpy
- $\Delta H = m C_p \Delta T$
- $PV = nRT = n k_B N_A T$       Ideal gas law

# Thermodynamics

## Problem 1

$O_2$  gas goes through the cycle  $A \rightarrow B \rightarrow C \rightarrow A$  as shown in the figure. If the gas goes through one full cycle, which of the following is true for the gas?

- (a)  $\Delta U = \Delta H = 0$ ,  $\Delta S > 0$ ,  $W > 0$ ,  $Q < 0$
- (b)  $\Delta U = \Delta H = 0$ ,  $\Delta S > 0$ ,  $W < 0$ ,  $Q > 0$
- (c)  $\Delta U = \Delta H = \Delta S = 0$ ,  $W < 0$ ,  $Q > 0$
- (d)  $\Delta U = \Delta S = 0$ ,  $\Delta H > 0$ ,  $W > 0$ ,  $Q < 0$
- (e)  $0 < \Delta U < \Delta H$ ,  $\Delta S > 0$ ,  $W < 0$ ,  $Q > 0$



# Thermodynamics

## Problem 2

A container is separated into 2 equal volumes by a membrane. The right side of the membrane is filled with 1 mole of  $\text{N}_2$  gas, and the left side is filled with 1 mole of  $\text{CO}_2$  gas. The membrane is broken and the two gasses are allowed to mix. What is the change in entropy?

- (a) 0
- (b)  $2R \ln(2)$
- (c)  $R \ln(2)$
- (d)  $C_v \Delta T/T$
- (e)  $C_p \Delta T/T$

# Quantum Mechanics

- Schrodinger equation
- Uncertainty Principle
- Harmonic oscillator
  - $H = \frac{p^2}{2m} + \frac{1}{2} m \omega^2 = \hbar \omega (a^* a + \frac{1}{2})$
- Infinite Square Well
- Delta function potential
- Angular Momentum Operators
- Hydrogen (like) Atom
- 1st order perturbation
- So on ...

$$-\frac{\hbar^2}{2m} \nabla^2 \Psi + V \Psi = E \Psi$$

$$\sigma_A^2 \sigma_B^2 \geq \left( \frac{1}{2i} \langle [\hat{A}, \hat{B}] \rangle \right)^2$$

$$E = \hbar \omega (n + \frac{1}{2})$$

$$\Psi_n = \sqrt{\frac{2}{a}} \sin\left(\frac{n\pi}{a} x\right)$$

$$E_n = - \left[ \frac{\mu}{2\hbar^2} \left( \frac{Ze^2}{4\pi\epsilon_0} \right)^2 \right] \frac{1}{n^2}$$

# Quantum Mechanics

## Problem 1

A diatomic molecule is initially in the state  $\Psi(\theta, \phi) = (5Y_1^1 + 3Y_5^1 + 2Y_5^{-1})/(38)^{1/2}$ , where  $Y_l^m$  is the spherical harmonic. What is the probability of measuring  $m=1$ ?

- (a) 25/38
- (b) 3/38
- (c)  $8/(38)^{1/2}$
- (d) 34/38
- (e) 4/5

# Quantum Mechanics

## Problem 2

A molecule is in the state  $\Psi(\theta, \varphi) = (3/4\pi)^{1/2} \sin \theta \cos \varphi$ . What is the expectation value for the operator  $O = [L_+, L_-]$ ?

- (a) 0
- (b)  $\hbar^2$
- (c)  $2\hbar^2$
- (d)  $3/2 \hbar^2$
- (e)  $3\hbar^2$

# Quantum Mechanics

## Problem 3

The operator  $O = (\mathbf{a}^* + \mathbf{a})^2$  operates on the a wave quantum harmonic oscillator in state  $|n=4\rangle$ . What is the expectation value for this operator?

(a) 9

(b) 8

(c) 0

(d)  $2\pi$

(e) 4



# Quantum Mechanics

## Problem 4

Muonic hydrogen is when a muon instead of an electron is orbiting the proton. The muon is approximately 200 time heavier than the electron. What is the ground state energy of a muonic hydrogen?

- (a) -2448 eV
- (b) -13.6 eV
- (c) Muons decay, and therefore muonic hydrogen does not exist.
- (d) -2720 eV
- (e) 27200 eV

# Quantum Mechanics

## Problem 5

Two hydrogen atoms with spin  $\mathbf{S}_1$  and  $\mathbf{S}_2$  have the interaction Hamiltonian  $H = -a \mathbf{S}_1 \cdot \mathbf{S}_2$  where  $J > 0$ . The atoms are placed in a uniform magnetic field  $B$ . What is their ground state energy?

(a)  $-(J/2)[(S_1 + S_2)(S_1 + S_2 + 1) - S_1(S_1 + 1) - S_2(S_2 + 1)]$

(b)  $-\gamma B S_1 - \gamma B S_2$

(c)  $-\gamma B S_1 - \gamma B S_2 - (J/2)[(S_1 + S_2)(S_1 + S_2 + 1) - S_1(S_1 + 1) - S_2(S_2 + 1)]$

(d)  $-\gamma B S_1 - \gamma B S_2 - J[S_1(S_1 + 1) + S_2(S_2 + 1)]$

(e)  $-\gamma B S_1 - \gamma B S_2 - J S_1 S_2$

# Answers

## Thermo

1)C

2)B

## QM

1)D

2)A

3)A

4)A

5)C